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Chapter 18
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SECTION 18.1
RATES OF
REACTION (pages
541-547) This

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section explains what is meant by the rate of a chemical reaction.

It also uses collision theory to show how the rate of a chemical reaction is influenced by the reaction conditions.

Collision Theory
(pages 541–544) 1.

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~~REACTION RATES~~

~~AND EQUILIBRIUM~~

~~18~~

a state of balance
in which the rates
of the forward and
reverse reactions
are equal; no net
change in the
amount of
reactants and
products occurs in
the chemical

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18.2 Reaction
equilibrium
position the
relative
concentrations of
reactants and
products of a
reaction that has
reached
equilibrium;
indicates whether
the reactants or
products are
favored in the

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18 reversible reaction
(18.2)

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a reaction in which
the conversion of
reactants into
products and the
conversion of
products into
reactants occur

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18.2) chemical equilibrium. a state of balance in which the rates of the forward and reverse reactions are equal; no net change in the amount of reactants and products occurs in the chemical system (18.2) Le

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fast a reaction
occurs. 14: Rates

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Reactions -

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before, there are

three reaction

rates in this

reaction: k_1 , k_{-1} ,

and k_2 . The pre-

equilibrium

approximation uses

the rate constants

to solve for the

rate of the

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18 Reaction, indicating
how

Rates

Equilibrium D
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Reaction Rates And
Equilibrium. In

layman's terms,

equilibrium is

defined as a state
of balance due to

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10 Reaction Rates
equal reactions of
opposing forces,
and today we'll be
talking all about it
with regards to the
scientific study of
chemistry, focusing
on such topics as
reaction rates.

~~Chapter 18~~

~~Reaction Rates And~~

~~Equilibrium~~

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"Reaction Rates
and Equilibrium"
Name: _____ 1.

Energy that is available to do work is called free energy. 2. Reaction rate is defined as the number of atoms, ions, or molecules that react in a given time to form

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products. 3.

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“Reaction Rates
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Cotton . Activation
Energy is being

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supplied Activated
Complex Read
slides 1-28, Stop at
Equilibrium
Constants

Answers

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Standardized Test
Prep - Page 643: 9.

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10 Reaction Rates
Answers
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Answer: True. Work
Step by Step. I. A
large value for an
equilibrium
constant indicates
that products are
favored at
equilibrium. True
(K_{eq} = products
over reactants so
as products
increase, K_{eq}
increases) Update
this answer!

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Rates of Reaction.
Collision Theory o
Rate = The speed
of any change that
occurs within an
interval of time o

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KEY = In chemistry, the rate of chemical change or the reaction rate is usually expressed as the amount of reactant changing per unit time o Collision Theory = atoms, ions, and molecules can react if they collide with one another, provided that the

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colliding particles
have enough
kinetic energy 1) If
the colliding
particles...

Answers

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Reaction Rates and
Equilibrium~~

Chapter 18
Reaction Rates And
Equilibrium. In
layman's terms,
equilibrium is

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18 defined as a state of balance due to equal reactions of opposing forces, and today we'll be talking all about it with regards to the scientific study of chemistry, focusing on such topics as reaction rates.

Chapter 18

Reaction Rates And Equilibrium -

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Rates of Reaction -
18.1 Lesson Check
- Page 601: 2

Answer The rate of
a chemical reaction
is dependent on

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10 Reaction
temperature,
concentration,
particle size, and
the use of a
catalyst.

Answers

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Rates ...~~

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08,11,12,13,15,16,30

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Answer Change in

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pressure, change in temperature, and change in concentration of reactants or products may disrupt a chemical system's equilibrium.

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~~Equilibrium 18.3~~

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"Reaction Rates and Equilibrium"

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reaction rate: the
number of particles
that react in a
given time to form
products: Le
Chatelier's
principle: If a stress

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18 Applied to a
system in dynamic
equilibrium, the
system changes to
relieve the stress
Answers

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the rates of the
forward or reverse
reactions are
equal, the reaction~~

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has reached a state of balance. indicates whether the reactants or products are favored in a reversible reaction. if a stress is applied to a system in dynamic equilibrium, the system changes in ways that relieves the stress.

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Reaction Rates and
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the rate of a
chemical change
expressed? in
chemistry, the rate
of chemical change
or the reaction rate

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